## 9<sup>TH</sup> CLASS

# Suspensions

Suspensions are heterogeneous mixtures in which solid particles are dispersed in a liquid or gas medium. Here are key points to understand about suspensions in chemistry:

### **Characteristics of Suspensions:**

#### 1. Heterogeneous Nature:

• Unlike solutions, suspensions are non-uniform and contain visibly distinguishable solid particles dispersed in a liquid or gas medium.

#### 2. Particle Size:

• Suspended particles are typically larger than those in colloids and solutions. They can range from a few micrometers to visible sizes.

#### 3. Settling:

• Due to gravity, suspended particles tend to settle at the bottom over time. This settling can be rapid or slow depending on the size and density of the particles.

### 4. Separation:

• Unlike colloids, where particles do not settle, suspensions can be separated by allowing the particles to settle or by using mechanical means.

### **\*** Examples of Suspensions:

### 5. Muddy Water:

• Natural water sources often contain suspended soil particles, giving the water a cloudy appearance.

### 6. Orange Juice with Pulp:

- Fruit juices with pulp contain suspended solid particles (pulp) that settle over time.
- 7. **Paint:** 
  - Paint is a suspension of pigment particles in a liquid medium. It needs to be stirred or shaken before use to redistribute the pigment.

### 8. Medical Suspensions:

- Some pharmaceutical formulations, such as certain liquid antibiotics or antacids, are suspensions.
- 9. Dust in the Air:

• Particles in the air, such as dust or pollen, can be considered suspended in the gas medium.

### Properties and Behavior:

### 10. Tyndall Effect:

• Similar to colloids, suspensions may exhibit the Tyndall effect, where the scattered light is visible when a beam of light passes through the mixture.

### 11. Stability:

• Suspensions are generally less stable than solutions and colloids. They require constant agitation or stirring to maintain a relatively uniform appearance.

### 12. Separation Methods:

• Filtration is a common method for separating solid particles from a liquid suspension. Sedimentation and decantation can also be used.

### 13. Re-suspension:

• Suspensions can be re-suspended by mechanical means, such as shaking or stirring, but they will eventually settle again due to gravity.

## Industrial Applications:

### 14. Pharmaceuticals:

• Some oral medications are formulated as suspensions to improve their palatability and ease of administration.

### 15. Paint Industry:

• Pigments in paint formulations are often in the form of suspended particles.

### 16. Environmental Engineering:

• In water treatment processes, the removal of suspended particles is a crucial step.

### 17. Food Industry:

• Some food products, like certain salad dressings, may be suspensions.